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APPLICATION NO.	FILING DATE	FIRST NAMED INVENTOR	ATTORNEY DOCKET NO.	CONFIRMATION NO.
09/965,617	09/27/2001	Trent M. Molter	PES-0039	8137
23462	7590	10/14/2003	EXAMINER	
CANTOR COLBURN, LLP 55 GRIFFIN ROAD SOUTH BLOOMFIELD, CT 06002				WILLS, MONIQUE M
		ART UNIT		PAPER NUMBER
		1746		

DATE MAILED: 10/14/2003

Please find below and/or attached an Office communication concerning this application or proceeding.

<b>Office Action Summary</b>	Application No.	Applicant(s)
	09/965,617	MOLTER ET AL.
	Examiner Wills M Monique	Art Unit 1746

-- The MAILING DATE of this communication appears on the cover sheet with the correspondence address --  
Period for Reply

A SHORTENED STATUTORY PERIOD FOR REPLY IS SET TO EXPIRE 3 MONTH(S) FROM  
THE MAILING DATE OF THIS COMMUNICATION.

- Extensions of time may be available under the provisions of 37 CFR 1.136(a). In no event, however, may a reply be timely filed after SIX (6) MONTHS from the mailing date of this communication.
- If the period for reply specified above is less than thirty (30) days, a reply within the statutory minimum of thirty (30) days will be considered timely.
- If NO period for reply is specified above, the maximum statutory period will apply and will expire SIX (6) MONTHS from the mailing date of this communication.
- Failure to reply within the set or extended period for reply will, by statute, cause the application to become ABANDONED (35 U.S.C. § 133).
- Any reply received by the Office later than three months after the mailing date of this communication, even if timely filed, may reduce any earned patent term adjustment. See 37 CFR 1.704(b).

**Status**

1) Responsive to communication(s) filed on 27 September 2001.

2a) This action is **FINAL**.      2b) This action is non-final.

3) Since this application is in condition for allowance except for formal matters, prosecution as to the merits is closed in accordance with the practice under *Ex parte Quayle*, 1935 C.D. 11, 453 O.G. 213.

**Disposition of Claims**

4) Claim(s) 1-35 is/are pending in the application.

4a) Of the above claim(s) \_\_\_\_\_ is/are withdrawn from consideration.

5) Claim(s) \_\_\_\_\_ is/are allowed.

6) Claim(s) 1-35 is/are rejected.

7) Claim(s) \_\_\_\_\_ is/are objected to.

8) Claim(s) \_\_\_\_\_ are subject to restriction and/or election requirement.

**Application Papers**

9) The specification is objected to by the Examiner.

10) The drawing(s) filed on \_\_\_\_\_ is/are: a) accepted or b) objected to by the Examiner.

    Applicant may not request that any objection to the drawing(s) be held in abeyance. See 37 CFR 1.85(a).

11) The proposed drawing correction filed on \_\_\_\_\_ is: a) approved b) disapproved by the Examiner.

    If approved, corrected drawings are required in reply to this Office action.

12) The oath or declaration is objected to by the Examiner.

**Priority under 35 U.S.C. §§ 119 and 120**

13) Acknowledgment is made of a claim for foreign priority under 35 U.S.C. § 119(a)-(d) or (f).

    a) All    b) Some \*    c) None of:

    1. Certified copies of the priority documents have been received.

    2. Certified copies of the priority documents have been received in Application No. \_\_\_\_\_.

    3. Copies of the certified copies of the priority documents have been received in this National Stage application from the International Bureau (PCT Rule 17.2(a)).

    \* See the attached detailed Office action for a list of the certified copies not received.

14) Acknowledgment is made of a claim for domestic priority under 35 U.S.C. § 119(e) (to a provisional application).

    a) The translation of the foreign language provisional application has been received.

15) Acknowledgment is made of a claim for domestic priority under 35 U.S.C. §§ 120 and/or 121.

**Attachment(s)**

1) Notice of References Cited (PTO-892)

2) Notice of Draftsperson's Patent Drawing Review (PTO-948)

3) Information Disclosure Statement(s) (PTO-1449) Paper No(s) \_\_\_\_\_

4) Interview Summary (PTO-413) Paper No(s) \_\_\_\_\_

5) Notice of Informal Patent Application (PTO-152)

6) Other: \_\_\_\_\_

**DETAILED ACTION**

**Information Disclosure Statement**

The information disclosure statement(s) filed March 21, 2002, August 30, 2002 and November 7, 2002 has/have been received and complies with the provisions of 37 CFR 1.97, 1.98 and MPEP § 609.

**International Search Report**

The U.S. Patent 5,432,023 cited as an "X" on the International Search Report does not when taken alone teach the instant invention. The reference teaches a catalyst but is silent to an integrated proton conductive material integrated with the catalyst at the range of up to about 50wt%.

**Claim Rejections - 35 USC § 102**

The following is a quotation of the appropriate paragraphs of 35 U.S.C. 102 that form the basis for the rejections under this section made in this Office action:

A person shall be entitled to a patent unless –

(b) the invention was patented or described in a printed publication in this or a foreign country or in public use or on sale in this country, more than one year prior to the date of application for patent in the United States.

Claims 21-22 & 24 are rejected under 35 U.S.C. 102(b) as being anticipated by Hayashi et al. GB 1,252,463.

Hayashi teaches making an air electrode by mixing an active carbon support with a silver catalyst and water repellent (page 3, lines 34-40). The carbon support is inherently non-oxidizable at anodic potentials of less than about 4 volts. Therefore, the instant claims are anticipated by Hayashi.

### **Claim Rejections - 35 USC § 102**

The following is a quotation of the appropriate paragraphs of 35 U.S.C. 102 that form the basis for the rejections under this section made in this Office action:

A person shall be entitled to a patent unless -

(b) the invention was patented or described in a printed publication in this or a foreign country or in public use or on sale in this country, more than one year prior to the date of application for patent in the United States.

Claims 21-22 & 24 are rejected under 35 U.S.C. 102(b) as being anticipated by Shimamune et al. EO 0 606 051.

Shimamune teaches making a gas electrode by mixing an active carbon support with a gold catalyst (abstract and Example 2``). The carbon support is inherently non-oxidizable at anodic potentials of less than about 4 volts. Therefore, the instant claims are anticipated by Shimanune.

### **Claim Rejections - 35 USC § 102**

The following is a quotation of the appropriate paragraphs of 35 U.S.C. 102 that form the basis for the rejections under this section made in this Office action:

A person shall be entitled to a patent unless –

(b) the invention was patented or described in a printed publication in this or a foreign country or in public use or on sale in this country, more than one year prior to the date of application for patent in the United States.

Claim 24 is rejected under 35 U.S.C. 102(b) as being anticipated by Tatchev WO 99/16546.

Tatchev teaches making a gas electrode by impregnating a support with gold (page 3 paragraph 5). Therefore, the instant claims are anticipated by Tatchev.

### **Claim Rejections - 35 USC § 102**

The following is a quotation of the appropriate paragraphs of 35 U.S.C. 102 that form the basis for the rejections under this section made in this Office action:

A person shall be entitled to a patent unless –

(e) the invention was described in (1) an application for patent, published under section 122(b), by another filed in the United States before the invention by the applicant for patent or (2) a patent granted on an application for patent by another filed in the United States before the invention by the applicant for patent, except that an international application filed under the treaty defined in section 351(a) shall have the effects for purposes of this subsection of an application filed in the United States only if the international application designated the United States and was published under Article 21(2) of such treaty in the English language.

Claims 1-32 are rejected under 35 U.S.C. 102(e) as being anticipated by Koschany et al. U.S. Patent 6,183,898.

Koschany teaches a gas diffusion electrode comprising a catalytically active material (col. 4, lines 50-55). The catalytically active material includes a catalyst from transition group VIII and IV and preferably platinum (col. 4, lines 49-55, claim 18) present in an amount greater than 2% (col. 4, lines 55-60) and an ion conducting polymer selected from perfluorinated polymers (fluorinated binder material such as perfluorinated polymers (fluorinated ethylene-propylene copolymers or polytetrafluoroethylene), polyether ketones, polyether sulfones, polysulfones, polybenzimidazoles, polyphenylene sulfides, polyimide, polyamide or polyphenylene oxides (col. 3, lines 15-25,) and a sulfonated fluorocarbon ion exchange resin such as Nafion® (col. 8, lines 50-55, claims 6-9) present in amount up to 90% (col. 4, lines 60-68) and electrically conductive carbon up to 50% (col. 3, lines 30-35, claims 11-15). The support material may include carbon or carbon fibers, glass fibers or fibers comprising organic polymers, for example polypropylene, polyester (polyethylene terephthalate), polyphenylene sulfide or poly ether ketons (col. 2, lines 40-50). The support is inherently non-oxidizable at anodic potentials of less than about 4 volts (claim 1, 10, 19) and a resistivity of less than about 270 microohm-cenitmeter (clam 16 & 32). The support may also include conductive metal such as gold, silver or nickel (col. 2, lines 60-68 (claims 11-15). The weight percent of the catalyst, carbon support and proton conductive material encompass the compositions in claims 1-5 & 27-31. The electrode is formed by mixing the support, catalyst and proton conductive material

(Example 1, claims 21,22 & 24-26. The catalytic active material may be employed in a fuel cell with a second electrode (col. 6, lines 55-65) and a proton exchange membrane interdisposed between (col. 5, lines 65-68 , claim 27). The catalytic active material may be coated with a polymer mesh of polyesters, polyether kentons, polyether sulfons, polysulfons (col. 6, lines 30-40, claims 23 & 26). Therefore, the instant claims are anticipated by Koschany.

### **Claim Rejections - 35 USC § 102**

The following is a quotation of the appropriate paragraphs of 35 U.S.C. 102 that form the basis for the rejections under this section made in this Office action:

A person shall be entitled to a patent unless -

(b) the invention was patented or described in a printed publication in this or a foreign country or in public use or on sale in this country, more than one year prior to the date of application for patent in the United States.

Claims 1, 4-14, 16-26, 27,30-32 are rejected under 35 U.S.C. 102(b) as being anticipated by Zuber et al. U.S. Patent 6,156,449.

Zuber teaches a catalyst layer for polymer electrolyte fuel cells comprising a catalyst layer on a substrate material, wherein the catalyst material contains a proton-conducting polymer (ionomer), electrically conductive carbon particles and fine particles of at least one precious metal (abstract). The ink contains no prefabricated support catalyst but has all the precursors necessary for the layer (solution of the ionomer, precious metal complex compounds and conductive

carbon particles), so that the catalyst layer can be produced in one operation. Subsequent introduction of the catalytically active precious metals is not necessary. Col. 5, lines 1-7, for claims 21,22,23 & 24. The precious metal catalyst includes platinum, palladium, iridium, rhodium and ruthenium or alloys thereof are used as the catalytically active components (col. 5, lines 8-13, claim 18). As ionomers for the ink, the materials of the proton-conducting membranes preferred ionmers include: tetrafluorothylene-fluorovinyl ether copolymers with acid functions, specially with sulfonic acid groups. Fluorine-free ionomer materials, such as sulfonated polyether ketones or aryl ketones or polybenzimidazoles are also suitable(col. 6, lines 5-15, claim 6) and a sulfonated fluorocarbon ion exchange resin such as Nafion® (Table 1, claims 6-9). The support material may include carbon particles, carbon blacks, graphite or activated carbon (col. 6, lines 20-25, claims 11-14 & 20). The precious metal catalyst may be present in an amount of 10 to 40 %, the carbon particles may be present in an amount of 60 to 90% and the ionomer may be present in an amount of 60 to 90% or less than 50% or (col. 6, lines 25-35 and Table 1 and Table 2, claims 1,4,5,27,30,31). The support is inherently non-oxidizable at anodic potentials of less than about 4 volts (claim 1, 10, 19) and resistivity of less than about 270 microohm-cenitmeter (clam 16 & 32). The electrode is employed in a fuel cell with a second electrode and a proton exchange membrane disposed in-between (col. 7, lines 40-55 claim 27). A proton exchange membrane is applied to the electrode catalyst (col. 7, lines 20-30, claims 23 & 26). Therefore, the instant claims are anticipated by Zuber.

### **Claim Rejections - 35 USC § 103**

The following is a quotation of 35 U.S.C. 103(a) which forms the basis for all obviousness rejections set forth in this Office action:

(a) A patent may not be obtained though the invention is not identically disclosed or described as set forth in section 102 of this title, if the differences between the subject matter sought to be patented and the prior art are such that the subject matter as a whole would have been obvious at the time the invention was made to a person having ordinary skill in the art to which said subject matter pertains. Patentability shall not be negated by the manner in which the invention was made.

Claim 33-35 are rejected under 35 U.S.C. 103(a) as being unpatentable over Dine et al. U.S. Pub. 2002/0098393, in view of Koschany et al. U.S. Patent 6,183,898.

Dine teaches an electrochemical cell comprising an anode 110 with an anode catalyst layer 112, a cathode 114 and a cathode catalyst layer 116, and a solid polymer membrane interposed in between (par. 27,28 & 9). The cathode flow field plate 120 is adjacent to the cathode and the anode flow field plate 118 is adjacent to the anode (par. 28). The anode flow field plate carries a hydrogen containing fuel across the anode from

an inlet 130 to an outlet 132 (par. 28). A water coolant loop 132 is in communication with one of the electrodes (par. 28 and /Fig. 1).

Dine is silent to an electrode catalyst material comprising the composition comprising 5 to 95 wt % of a support material, 5 to 95% of a catalyst and up to 50 wt% of a proton conductive material.

Koschany teaches a catalytically active material as described hereinabove that is mechanically stable, and has high electrical conductivity (col. 1, lines 57-68).

Therefore, it would have been obvious to one of ordinary skill in the art at the time the invention was made to employ the catalytic material of Koschany in the fuel cell of Dine, in order to increase the mechanically stability and electrical conductivity of the electrodes.

### **Claim Rejections - 35 USC § 103**

The following is a quotation of 35 U.S.C. 103(a) which forms the basis for all obviousness rejections set forth in this Office action:

(a) A patent may not be obtained though the invention is not identically disclosed or described as set forth in section 102 of this title, if the differences between the subject matter sought to be patented and the prior art are such that the subject matter as a whole would have been obvious at the time the invention was made to a person having ordinary skill in the art to which said subject matter pertains. Patentability shall not be negated by the manner in which the invention was made.

Claim 33-35 are rejected under 35 U.S.C. 103(a) as being unpatentable over Dine et al. U.S. Pub. 2002/0098393, in view of Zuber et al. U.S. Patent 6,156,449.

Dine teaches an electrochemical cell comprising an anode 110 with an anode catalyst layer 112, a cathode 114 and a cathode catalyst layer 116, and a solid polymer membrane interposed in between (par. 27,28 & 9). The cathode flow field plate 120 is adjacent to the cathode and the anode flow field plate 118 is adjacent to the anode (par. 28). The anode flow filed plate carries a hydrogen containing fuel across the anode from

an inlet 130 to an outlet 132 (par. 28). A water coolant loop 132 is in communication with one of the electrodes (par. 28 and /Fig. 1).

Dine is silent to an electrode catalyst material comprising the composition comprising 5 to 95 wt % of a support material, 5 to 95% of a catalyst and up to 50 wt% of a proton conductive material.

Zuber teaches a catalytically active material as described hereinabove that is increases electrical power of the electrode (col. 4, lines 45-50). Therefore, it would have been obvious to one of ordinary skill in the art at the time the invention was made to employ the catalytic material of Koschany in the fuel cell of Dine, in order to increase the electrical power of the electrode.

### **Conclusions**

The prior art made of record and not relied upon is considered pertinent to applicant's disclosure. Allen et al. U.S. Patent 5,958,197 teaches a catalyst for gas diffusion electrodes. Shinkai et al. U.S. Patent 6,187,468 teaches electrodes for fuel cells.

Any inquiry concerning this communication or earlier communications from the Examiner should be directed to Monique Wills whose telephone number is (703) 305-0073. The Examiner can normally be reached on Monday-Friday from 8:30am to 5:00 pm.

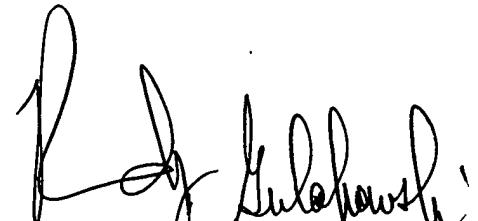
Any inquiry of a general nature or relating to the status of this application or proceeding should be directed to the Group receptionist whose telephone number is (703) 308-0661.

If attempts to reach Examiner by telephone are unsuccessful, the Examiner's supervisor, Randy Gulakowski, may be reached at 703-308-4333.

The unofficial fax number is (703) 305-3599. The Official fax number for non-final amendments is 703-872-9310. The Official fax number for after final amendments is 703-872-9311.

Mw

09/27/03



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